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Contents of dissertation abstract of doctor of philosophy (PhD) on technical sciences

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Sadikov Jahongir Nasirdjanovich

Research and development of technology for the manufacture of electrodes based on local raw materials for wear-resistant surfacing..... 31

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(PhD)

Фалсафа доктори (PhD) диссертацияси мавзуси Ўзбекистон Республикаси Вазирлар Маҳкамаси ҳузуридаги Олий аттестация комиссиясида В2020.2.PhD/T1589 рақам билан рўйхатга олинган.

Диссертация Тошкент давлат техника университетида бажарилган.

Диссертация автореферати уч тилда (ўзбек, рус тили ва инглиз тилида (резюме)) Илмий кенгашнинг веб-саҳифасида (www.tdtu.uz) ва «Ziynet» Ахборот таълим порталида (www.ziynet.uz) жойлаштирилган.

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Етакчи ташкилот: **Андижон машиносозлик институти**

Диссертация ҳимояси Тошкент давлат техника университети ҳузуридаги DSc.03/30.12.2019.T.03.04. рақамли Илмий кенгашнинг 2022 йил «26» феврал соат 14⁰⁰ даги мажлисида бўлиб ўтади. (Манзил: 100095, Тошкент шаҳри, Университет кўчаси, 2-уй. Тел./факс:(99871)227-10-32, e-mail: tadqiqotchi@tdtu.uz)

Диссертация билан Тошкент давлат техника университети Ахборот-ресурс марказида танишиш мумкин (242 - рақами билан рўйхатга олинган). (Манзил: 100095, Тошкент шаҳри, Университет кўчаси, 2-уй. Тел.: (99871)227-10-32.)

Диссертация автореферати 2022 йил «11» феврал куни тарқатилди.
(2022 йил «11» февралдаги 134 - рақамли реестр баённомаси).



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К.А.Каримов

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η_{E_z}

$$\bar{\eta}_{E_z} = a\eta_{E_z}^{\acute{o}\acute{c}\hat{a}\hat{e}} + b\eta_{E_z}^{\acute{o}\hat{d}\hat{d}} + c\eta_{E_z}^{\acute{o}\acute{e}\cdot\hat{i}\hat{e}\hat{i}} + d\eta_{E_z}^{\acute{o}\acute{e}\cdot\hat{n}\acute{o}\hat{i}\cdot\hat{i}\hat{i}\hat{a}} \quad (1)$$

$$a + b + c + d = 1 \quad (2)$$

$$\bar{\eta}_{E_z} = \frac{m_{E_z}^{\acute{y}\acute{d}\cdot\hat{e}\hat{i}\hat{i}}}{m_{E_z}^{\acute{a}\hat{d}\hat{a}\hat{e}\hat{a}\hat{\phi}}} \quad (3)$$

$m_{E_z}^{\acute{a}\hat{d}\hat{a}\hat{e}\hat{a}\hat{\phi}}$

$$\eta_{E_z} = 1 - k_1^{E_z} \quad (4)$$

$$\eta_{E_z} = 1 - k_2^{E_z} \quad (5)$$

$$\eta_{E_z} = k_3^{E_z} \quad (6)$$

$$\eta_{E_z} = k_4^{E_z} \quad (7)$$

$\eta_{E_z}, \eta_{E_z}, \eta_{E_z}$

$k_1^{E_z}$

$$k_3^{E_z} - (1 - k_4^{E_z}) E_z ;$$

$$k_5^{E_z} - (3 - E_z)$$

2.

$$m_{E_z} = m^c \cdot \frac{E_z^c}{100} + \sum_{k=1}^l m_k \cdot \frac{E_z}{100} + \sum_{k=1}^p m_k \cdot \frac{E_z}{100} + \sum_{k=1}^s m_k \cdot \frac{E_z}{100} = m_{E_z} + m_{E_z} + m_{E_z} + m_{E_z}$$

$m_{E_z}, m_{E_z}, m_{E_z}$

2.1.

$$m_{E_z} = \frac{m}{1+k} \cdot \frac{E_z}{100}, \quad (9)$$

k

$$k = \frac{m}{m} \quad (10)$$

$$m = \frac{\pi}{2} (D^2 - d^2) L \gamma - L$$

$\gamma -$, $/^3$, $m = \frac{\pi d^2}{4} L \gamma -$

L

$;$ $\gamma -$

$/^3$

2.2.

E_z

$$m_{E_z} = \frac{k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} m \sum_{k=1}^l \frac{\%}{100} \cdot \frac{[E_z]_k}{100} \quad (11)$$

$$m_{E_z}^k = m \cdot \frac{\%}{100} \quad (12)$$

$$m_{\hat{e}\ddot{u}} = \frac{k_{\hat{e}\ddot{u}}}{(1+k_{\hat{e}\ddot{u}}) \cdot (1+0,01\alpha \cdot \beta)} m_{\hat{y}\ddot{e}} \quad (13)$$

, %;

$$\beta = 0,59\gamma_{\bar{n}\phi} + 0,028m - 0,535, \quad (14)$$

$\gamma_{\bar{n}\phi}$ ó $\gamma_{\bar{n}\phi} = \frac{\gamma_{\bar{n}\phi}}{\gamma_{\bar{n}\phi} + \gamma_{\bar{n}\phi}}$;

m ó
2.3.

E_z

$$m_{E_z} = m \cdot j + m \quad (15)$$

2.3.1.

$$m_{\hat{A}_y}^{\phi\bar{e} \cdot i\bar{e} j} = \frac{m_{\bar{y}\bar{e}} \cdot k_{\bar{e}i}}{(1+k_{\bar{e}i}) \cdot (1+0,01\alpha \cdot \beta)} \cdot \sum_{k=1}^z \frac{\% \phi\bar{e} \cdot i\bar{e} j}{100} \cdot \frac{(E_{z_n} O_m)_j}{100} \cdot \frac{M_{\hat{A}_z}}{M_{E_{z_n} O_m}} \quad (16)$$

2.3.2.

$$m_{E_z} = \frac{m \cdot k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} \cdot \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j}{100} \cdot \frac{M}{M_{E_{z_n} O_m}} \quad (17)$$

$$m_{E_z} = \frac{m \cdot k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} \cdot \frac{M}{M_{E_{z_n} O_m}} \left(\sum_{k=1}^p \frac{\% \cdot j}{100} \cdot \frac{(E_{z_n} O_m)_j}{100} + \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j}{100} \right) \quad (18)$$

$$m_{E_z} = \frac{m}{1+k} \cdot \left(\frac{E_z^c}{100} (1-k_1^{E_z}) + \frac{k}{(1+0,01\alpha \cdot \beta)} \left(m \sum_{k=1}^l \frac{\% \cdot k}{100} \cdot \frac{[E_z]_k}{100} (1-k_2^{E_z}) + \frac{M}{M_{E_{z_n} O_m}} (k_3^{E_z} + k_4^{E_z}) \left(\sum_{k=1}^p \frac{\% \cdot j}{100} \cdot \frac{(E_{z_n} O_m)_j}{100} + \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j}{100} \right) \right) \right) \quad (19)$$

3.

E_z

$$\bar{\eta}_{E_z} = \frac{[E_z]}{a[E_z] + b[E_z] + c[E_z] + d[E_z]} \quad (20)$$

$$\hat{a} = \frac{m_{\bar{y}\bar{e}} \cdot (1-k_{\bar{n}\phi}^{i\bar{a}})}{m_{i\bar{a}\phi} \cdot \hat{a}i \cdot (1+k_{\bar{e}i})}; \quad (21)$$

$$b = \frac{m \cdot (1-k) \cdot k \cdot \sum_{k=1}^l \frac{\% \cdot k}{100}}{m \cdot (1+k) \cdot \left(1 + \frac{\alpha\beta}{100} \right)}; \quad (22)$$

$$d = \frac{m \cdot (1 - k) \cdot k \cdot \sum_{k=1}^s \frac{\%}{100} \cdot k}{m \cdot (1 + k) \cdot \left(1 + \frac{\alpha\beta}{100}\right)} \quad (23)$$

$$c = 1 - a - b - d \quad (24)$$

$[E_z]$ - E_z

, %;

$[E_z]$ - E_z

, %;

$[E_z]$ - E_z

, %;

$[E_z]$ - $E_z \cdot O_m$

, %;

$[E_z]_{\text{н.о.и.}} - E_z$

, %.

4. :

$$m_{E_z} = \frac{m \cdot [E_z]}{100} \quad (25)$$

$m_{\text{ид.д.}} \text{ ó}$, ; $[E_z]$ ó

5. , %.

E_z

$$\eta_{E_z} = \frac{[E_z]}{[E_z] \cdot \gamma + [E_z] \cdot \gamma} \quad (26)$$

(1) - (26) -590 -

(6 ,

$k = 0,62$)

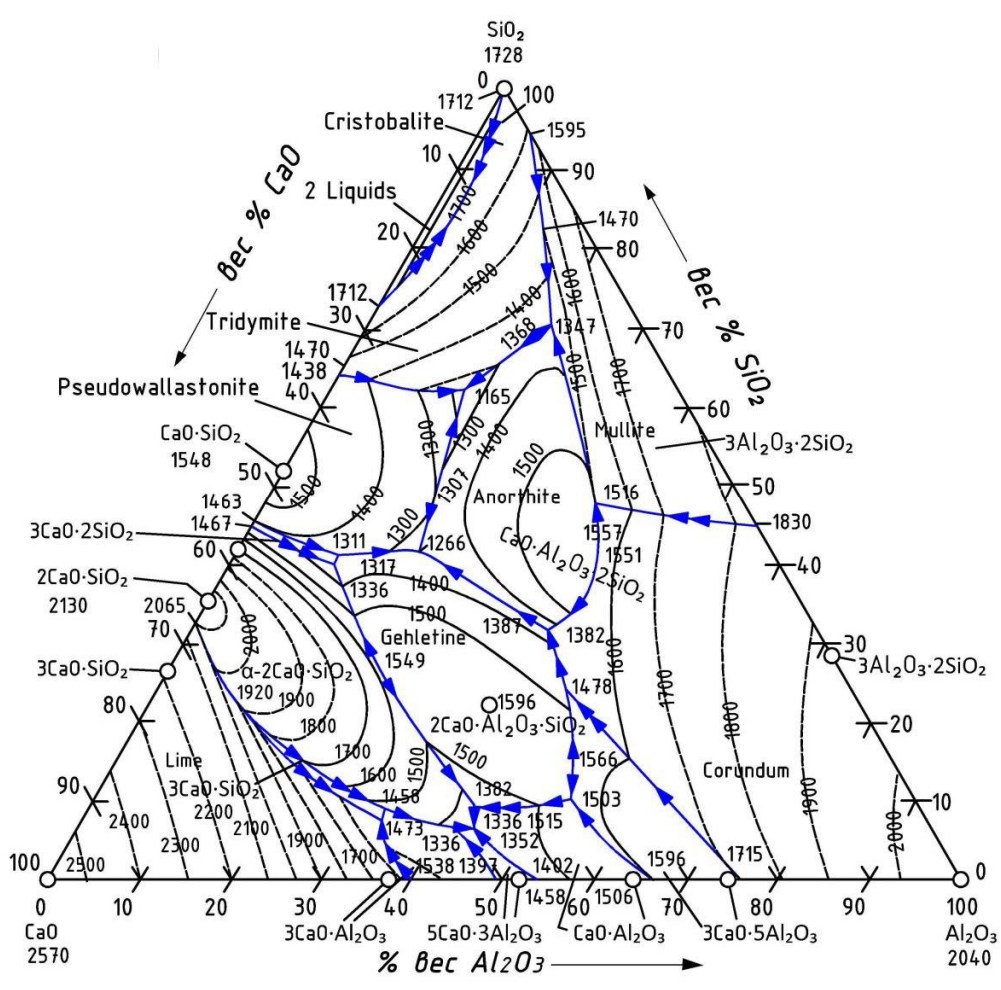
1-

	η	η_{Si}	η_{Mn}	η_i	η_{Cr}
	0,73	0,45	0,89	0,04	0,02
	0,72	0,44	0,88	0,04	0,02

5%

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1- $\text{CaO-SiO}_2\text{-Al}_2\text{O}_3$

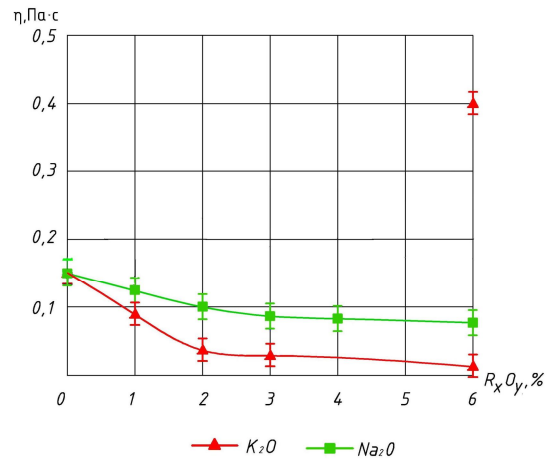
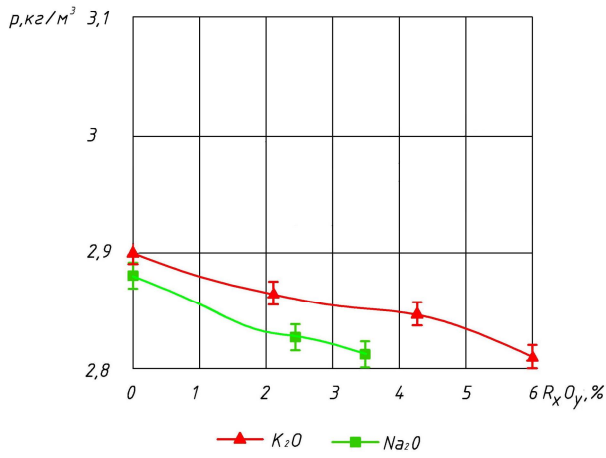
C $\text{CaO-SiO}_2\text{-Al}_2\text{O}_3$ 1300°C ~ 50% SiO_2 , ~ 30%
 ~ 20% Al_2O_3

: 1350°C C $\cdot\text{Al}_2\text{O}_3\cdot\text{SiO}_2$
 20% C , 37% Al_2O_3 43% SiO_2 1390°
 2C $\cdot\text{Al}_2\text{O}_3\cdot\text{SiO}_2$ 41% C , 37% Al_2O_3 22% SiO_2
 1300°C
 C $\text{-SiO}_2\text{-Al}_2\text{O}_3$

48-54% C

$\text{CaO-SiO}_2\text{-Al}_2\text{O}_3$

(,) Na_2O K_2O (,)
 7,08% 4,15%)



2- . CaO-SiO₂-Al₂O₃

3- . CaO-SiO₂-Al₂O₃

K₂O Na₂O

Na₂O K₂O

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Na₂O K₂O

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Na₂O²

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Na₂O K₂O

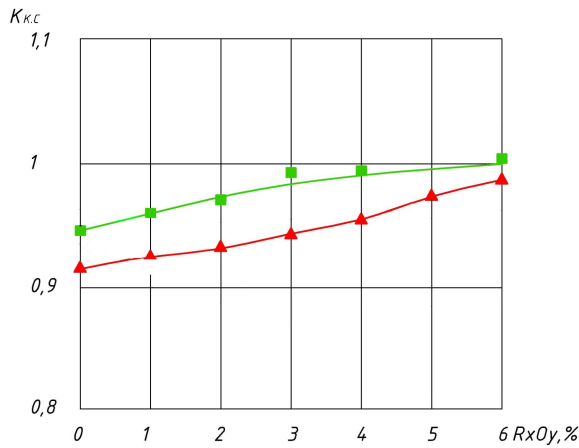
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, Na₂O K₂O

K₂O

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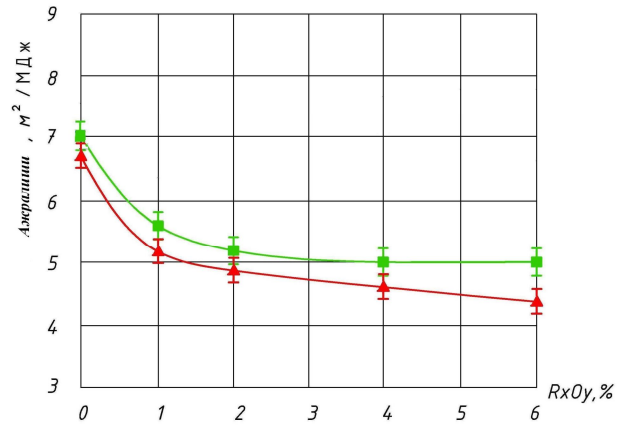
Na₂O



■ - Na₂O ▲ - K₂O

4- . CaO -SiO₂- Al₂O₃

Na₂O K₂O



■ - Na₂O ▲ - K₂O

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CaO -SiO₂- Al₂O₃
Na₂O K₂O

, Na₂O K₂O

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- 8-10; () - 8-10,

- 26-30; - 8-10, () -1-3;

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() - 5-6; () - 4-6, - 2-4,

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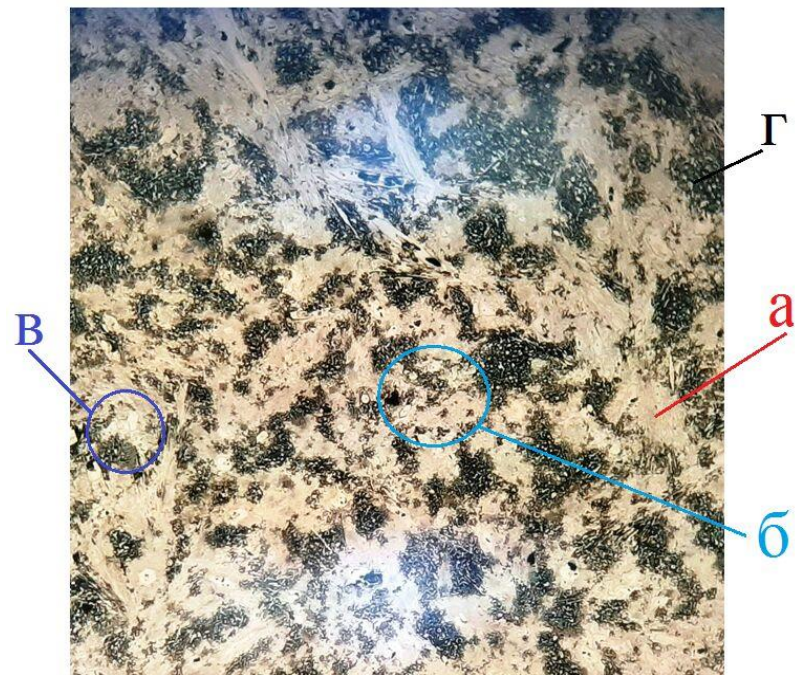
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- 26-28; - 8-10, () -1-3.

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						HRC
	1-	2-	3-	4-		
1	9,07	9,16	9,12	9,19	9,14	57-61
2	7,69	7,85	7,72	7,93	7,78	61-63
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4	8,53	8,86	8,94	8,56	8,70	59-62

15% (2-).

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4. CaO -SiO₂- Al₂O₃

. Na₂O K₂O Na₂O K₂O

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(PhD)

Тема диссертации доктора философии (PhD) по техническим наукам зарегистрирована в Высшей аттестационной комиссии при Кабинете Министров Республики Узбекистан за B2020.2.PhD/T1589.

Диссертация выполнена в Ташкентском государственном техническом университете. Автореферат диссертации на двух языках (узбекский, русский и английский (резюме)) размещен на веб-странице (www.tdtu.uz) и информационно-образовательном портале «Ziyonet» (www.ziyonet.uz).

Научный руководитель: Эрматов Зиядулла Досматович
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Ведущая организация: Андижанский машиностроительный институт

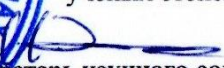
Защита диссертации состоится «26» февраля 2022 года в 14⁰⁰ часов на заседании Научного совета DSc.03/30.12.2019.T.03.04. при Ташкентском государственном техническом университете. (Адрес: 100095, г.Ташкент, ул. Университетская, 2. Тел./ факс:(99871)227-10-32, e-mail: tadqiqotchit@tdtu.uz)

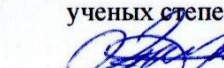
С диссертацией можно ознакомиться в Информационно-ресурсном центре Ташкентского государственного технического университета (зарегистрирована за № 242). (Адрес: 100095, г. Ташкент, ул. Университетская, 2. Тел.:(99871)227-10-32.)

Автореферат диссертации разослан «11» февраля 2022 года.
(реестр протокола рассылки №134 от «11» февраля 2022 года).




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по присуждению ученых степеней, д.т.н., профессор

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 2018 « 2017-2019 », -3682 27
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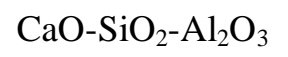
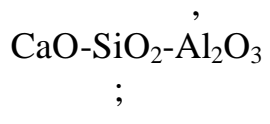
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Al_2O_3

CaO-SiO_2



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13-15%;

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CaO-SiO₂-Al₂O₃,
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 - E_z , -
 - ; $\bar{\eta}_{E_z}$

1. E_z :

$$\bar{\eta}_{E_z} = a\eta_{E_z} + b\eta_{E_z} + c\eta_{E_z} + d\eta_{E_z} \quad (1)$$

$$a+b+c+d=1 \quad (2)$$

$\bar{\eta}_{E_z}$ -
 E_z , :

$$\bar{\eta}_{E_z} = \frac{m_{E_z}}{m_{E_z}}, \quad (3)$$

m_{E_z} -
 , ;
 m_{E_z} - E_z , a,

b, c d e - , ,

$$\eta_{E_z} = 1 - k_1^{E_z}, \quad (4)$$

$$\eta_{E_z} = 1 - k_2^{E_z}, \quad (5)$$

$$\eta_{E_z} = k_3^{E_z}, \quad (6)$$

$$\eta_{E_z} = k_4^{E_z}, \quad (7)$$

η_{E_z} , η_{E_z} , η_{E_z} η_{E_z} -
 E_z , , ,

$k_1^{E_z}$ - E_z ,
 ;

$k_3^{E_z}$ - E_z
 (1.), ;

$k_4^{E_z}$ - E_z
 (2.),
 ;

$k_5^{E_z}$ - E_z
 (3.),

2. .

$$m_{E_z} = m^c \cdot \frac{E_z^c}{100} + \sum_{k=1}^l m_k \cdot \frac{E_z}{100} + \sum_{k=1}^p m_k \cdot \frac{E_z}{100} + \sum_{k=1}^s m_k \cdot \frac{E_z}{100} = m_{E_z} + m_{E_z} + m_{E_z} + m_{E_z} \quad (8)$$

2.1.

$$m_{E_z} = \frac{m}{1+k} \cdot \frac{E_z^c}{100}, \quad (9)$$

k

$$k = \frac{m}{m_c} \quad (10)$$

$$m = \frac{\pi}{2}(D^2 - d^2)L\gamma - L, \text{ d ó}$$

, D ó, γ -

$$, / ^3. m_c = \frac{\pi d^2}{4} L\gamma_c - L;$$

γ_c -

2.2.

E_z,

$$m_{E_z} = \frac{k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} m \sum_{k=1}^l \frac{\%}{100} \cdot \frac{[E_z]_k}{100} \quad (11)$$

$$m_{E_z}^k = m \cdot \frac{\%}{100} \quad (12)$$

$$m = \frac{k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} m, \quad (13)$$

.%;

$$\beta = 0,59\gamma + 0,028m - 0,535, \quad (14)$$

γ ó, / ^3;

m ó

2.3.

E_z,

$$m_{E_z} = m \cdot j + m \quad (15)$$

2.3.1.

E_z,

$$m \cdot j = \frac{m \cdot k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} \cdot \sum_{k=1}^p \frac{\% \cdot j \cdot (E_{z_n} O_m)_j \cdot M}{100 \cdot 100 \cdot M_{E_{z_n} O_m}} \quad (16)$$

2.3.2.

$E_z,$

:

$$m = \frac{m \cdot k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} \cdot \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j \cdot M}{100 \cdot 100 \cdot M_{E_{z_n} O_m}} \quad (17)$$

$$m_{E_z} = \frac{m \cdot k}{(1+k) \cdot (1+0,01\alpha \cdot \beta)} \cdot \frac{M}{M_{E_{z_n} O_m}} \left(\sum_{k=1}^p \frac{\% \cdot j \cdot (E_{z_n} O_m)_j}{100 \cdot 100} + \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j}{100 \cdot 100} \right) \quad (18)$$

E_z

:

$$m_{E_z} = \frac{m}{1+k} \cdot \left(\frac{E_z^c}{100} (1-k_1^{E_z}) + \frac{k}{(1+0,01\alpha \cdot \beta)} \left(m \sum_{k=1}^l \frac{\% \cdot k \cdot [E_z]_k}{100 \cdot 100} (1-k_2^{E_z}) + \frac{M}{M_{E_{z_n} O_m}} (k_3^{E_z} + k_4^{E_z}) \left(\sum_{k=1}^p \frac{\% \cdot j \cdot (E_{z_n} O_m)_j}{100 \cdot 100} + \frac{\alpha \cdot \beta \cdot (E_{z_n} O_m)_j}{100 \cdot 100} \right) \right) \right) \quad (19)$$

3.

$E_z:$

$$\bar{\eta}_{E_z} = \frac{[E_z]}{a[E_z] + b[E_z] + c[E_z] + d[E_z]}, \quad (20)$$

$$= \frac{m \cdot (1-k)}{m \cdot (1+k)}; \quad (21)$$

$$b = \frac{m \cdot (1-k) \cdot k \cdot \sum_{k=1}^l \frac{\% \cdot k}{100}}{m \cdot (1+k) \cdot \left(1 + \frac{\alpha\beta}{100}\right)}; \quad (22)$$

$$d = \frac{m \cdot (1-k) \cdot k \cdot \sum_{k=1}^s \frac{\% \cdot k}{100}}{m \cdot (1+k) \cdot \left(1 + \frac{\alpha\beta}{100}\right)} \quad (23)$$

$$c = 1 - a - b - d \quad (24)$$

$$\begin{aligned} [E_z] & - E_z, \% ; \\ [E_z] & - E_z, \% ; \\ [E_z] & - E_z, \% ; \\ [E_z] & - E_z, \% ; \end{aligned}$$

4. $[E_z]$... E_z ... %;

4.
:

$$m_{E_z} = \frac{m \cdot [E_z]}{100} \quad (25)$$

5. m ... E_z ... %.

$$\eta_{E_z} = \frac{[E_z]}{[E_z] \cdot \gamma + [E_z] \cdot \gamma} \quad (26)$$

T-590 (6, $k = 0,62$)

η_{E_z}

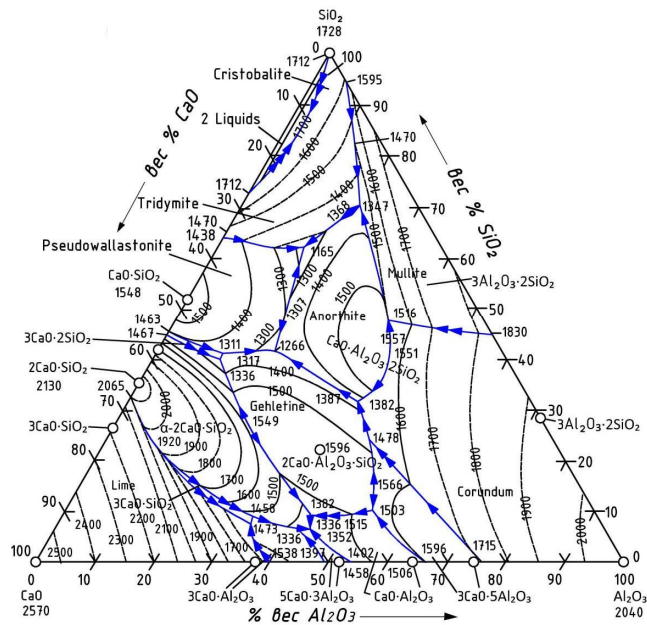
	η	η_{Si}	η_{Mn}	η_i	η_{Cr}
	0,73	0,45	0,89	0,04	0,02
	0,72	0,44	0,88	0,04	0,02

5 %,

«

»

~50% SiO_2 , ~30% Al_2O_3 , ~20% CaO , 1300°
 : $CaO \cdot Al_2O_3 \cdot SiO_2$, 20% CaO , 37% Al_2O_3 , 43% SiO_2
 1350°, $2CaO \cdot Al_2O_3 \cdot SiO_2$, 41% CaO ,
 37% Al_2O_3 , 22% SiO_2 , 1390°
 1300°, SiO_2 - Al_2O_3 ,
 48% 54%



. 1.

CaO-SiO₂-Al₂O₃

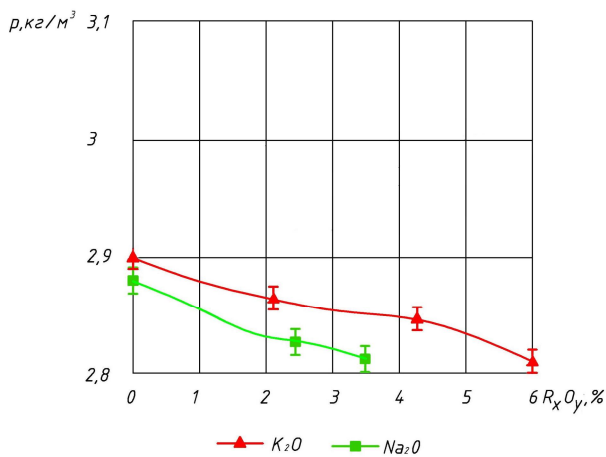
7,08% 4,15%
 Na₂O 2O ()
 Al₂O₃, (CaO-SiO₂-

. 2,

Na₂O 2O

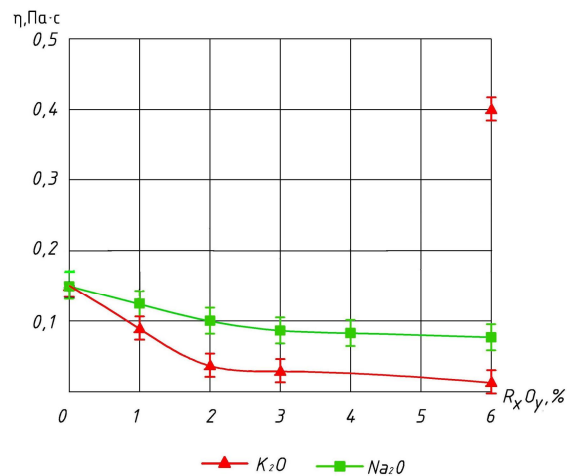
= 1700

. 3.



. 2.

Al₂O₃ CaO -SiO₂-
 Na₂O 2



. 3.

Al₂O₃ CaO -SiO₂-
 Na₂O 2O

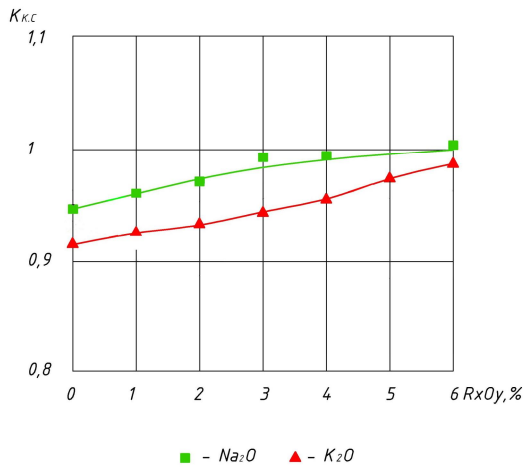
K₂O

Na₂O.

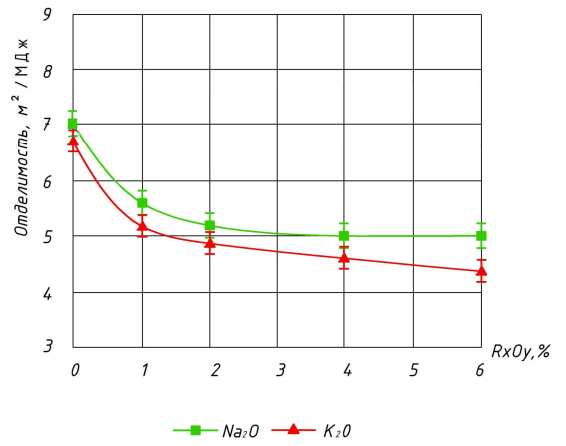
Na₂O K₂O . 4. Na₂O 2O . 4,

Na₂O 2O . 5.

Na₂O K₂O



. 4.



. 5.

CaO-SiO₂-Al₂O₃

Na₂O 2O 2O

Na₂O
CaO-SiO₂-Al₂O₃

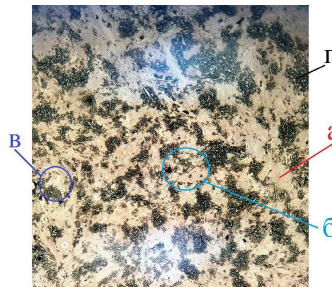
4³ 09 2

1 ó 590 (),
 2 ó . %: ó 8-10; () ó 10-12;
 () ó 3-5; () ó 5-7,
 ó 3-5, ó 8-10;
 ()- 8-10; ó 26-30; ó 8-10;
 () -1-3;

3 ó . %: ó 8-10, () ó 8-10;
 () ó 5-6; () ó 4-6,
 ó 2-4, ó 8-10; - 8-10
 (), ó 28-30; ó 9-11, ()
 -2-4;

4 ó . %: ó 8-10%; () ó 6-8;
 () ó 5-6; () ó 4-6;
 ó 2-4; (« »)- 8-10,
 ó 8-10; ó 26-28; ó 8-10, () -
 1-3.

2 ó ; ; ; ()
 (. 6).



. 6. ó
 , x200: ó ; ó

57-61 HRC. ,
 6-8 , 10-13 ,
 ó 8-10 , 5-6 .

1- 4

						HRC ,
	1	2	3	4		
1	9,07	9,16	9,12	9,19	9,14	57-61
2	7,69	7,85	7,72	7,93	7,78	61-63
3	7,89	7,96	7,94	8,02	7,95	61-63
4	8,53	8,86	8,94	8,56	8,70	59-62

590 , 2-4 - ,

2-4

13-15% (. 2).

.

(PhD)

«

»

:

1.

(, ,).

2.

8-10%

;

3.

(, ,).

4.

9-11%
Na₂O ₂O
CaO-SiO₂-Al₂O₃,
Na₂O ₂O

5.

8-10%;

6.

13-15%.

**SCIENTIFIC COUNCIL DSc.03/30.12.2019.T.03.04 ON THE
ADMISSION OF SCIENTIFIC DEGREES AT THE TASHKENT STATE
TECHNICAL UNIVERSITY**

TASHKENT STATE TECHNICAL UNIVERSITY

SADIKOV JAHONGIR NASIRDJANOVICH

**RESEARCH AND DEVELOPMENT OF TECHNOLOGU FOR
MANUFACTURE OF ELECTRODES BASED ON LOCAL RAW
MATERIALS FOR WEAR-RESISTANT SURFACING**

05.02.01 - Materials Science in Mechanical Engineering. Foundry. Heat treatment and treatment of metals by pressure. Metallurgy of ferrous, non-ferrous and rare metals. Technology of unique, rare and radioactive elements (in the direction of foundry production and metall processing technology)

ABSTRACT
of thesis of Doctor of Philosophy (PhD) in Technical Sciences

Tashkent - 2022

The theme of the dissertation of The Doctor of Philocophy (PhD) in technical sciences is registered in the Higher Attestation Commission under the Cabinet of Ministers of the Republic of Uzbekistan for B2020.2.PhD/T1589.

The thesis was performed at the Tashkent State Technical University.

The abstract of the thesis in two languages (Uzbek, Russian and English (summary)) is available on the website (www.tdtu.uz) and the information and educational portal "Ziyonet" (www.ziyonet.uz).

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Official opponents: **Abdullaev Fatkhulla Cagdullaevich**
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Xudoyarov Suleyman Rashidovich
candidate of technical science, associate professor

Lead organization: **Andijane mashine building institute**

Defense of the thesis will be held "26" february 2022 at 14⁰⁰ hours at a meeting of the Scientific Council DSc.03/30.12.2019.T.03.04. at the Tashkent State Technical University and the National University of Uzbekistan. (Address: 100095, Tashkent, Universitet str., 2. Tel./fax: (99871) 227-10-32, e-mail: tadqiqotchi@tdtu.uz)

The thesis is available at the Information and Resource Center of the Tashkent State Technical University (registered under No. 242). (Address: 100095, Tashkent, Universitet St., 2. Phone: (99871) 227-10-32.)

The thesis abstract was sent out «11» february 2022 year.
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INTRODUCTION (abstract of PhD thesis)

The aim of the research work is to develop a slag base and technology for the production of electrodes for wear-resistant surfacing based on local mineral raw materials.

The object of the research is the mineral raw materials of the Republic of Uzbekistan and the cast deposited metal formed as a result of manual arc surfacing with coated electrodes.

Scientific novelty of the research work is as follows:

the composition of the electrode coating for manual arc surfacing was developed based on the regularity of studying the chemical composition of silicon carbide, corundum, graphite, marble, pegmatite and quartz sand as possible components of welding materials, which increased the wear resistance of the deposited metal under conditions of abrasion by abrasive materials;

the composition of the slag base of surfacing electrodes from local ore-mineral materials was developed, selected according to the melting diagram of the CaO-SiO₂-Al₂O₃ slag system, which increased the strength characteristics of the deposited metal;

the composition of the gas-forming components of the coating of surfacing electrodes was developed, depending on the temperature of their dissociation, which made it possible to reduce the content of nitrogen, oxygen and hydrogen in the metal;

a mathematical model of the deposited metal formation process has been developed, depending on the thermophysical properties of the deoxidizing components in the composition of the electrode coating.

the influence of the introduction of sodium and potassium oxides on the basis of the physical and technological properties of the CaO-SiO₂-Al₂O₃ slag system of electrodes for wear-resistant surfacing was revealed.

The outline of the thesis. Based on the results obtained on the development of electrode coatings from local raw materials for wear-resistant surfacing:

the developed technology for the production of electrodes was introduced in JV LLC «Tashkent Pipe Plant» (reference 01/193-01-01 dated July 16, 2020, JSC «Uzmontazhspetsstroy») to optimize the content of deoxidizers in the composition of the coating of welding electrodes. As a result, the wear resistance of the deposited metal increased by 13-15% under the conditions of abrasion by abrasive materials;

an improved composition of the slag base of surfacing electrodes from local ore-mineral materials, selected according to the melting diagram of the CaO-SiO₂-Al₂O₃ slag system, was introduced into the JV LLC «Tashkent Pipe Plant» (reference 01/193-01-01 dated July 16, 2020, JSC «Uzmontazhspetsstroy»). As a result, the strength characteristics of welded joints have increased by 9-11%;

the proposed composition of the gas-forming components of the coating of surfacing electrodes was introduced in JV LLC «Tashkent Pipe Plant» (reference 01/193-01-01 dated July 16, 2020, JSC «Uzmontazhspetsstroy»). For this reason,

the content of nitrogen, oxygen and hydrogen in the weld metal decreased by 8-10%.

The structure and volume of the thesis. The thesis consists of an introduction, four chapters, conclusion, the list of used literature and appendixes. The thesis volume consists of 120 pages.

LIST OF PUBLISHED WORKS

I- (I- ; I-part)

1. Sadykov J.N., Ermatov Z.D. Mineral resources of the Republic of Uzbekistan for the production of electrode coatings of electrodes for wear-resistant// International Journal Of Advanced Research in Science, Engineering and Technology ó India, 2020. ó Vol.7, 5 (May). ó pp. 13891 ó 13894 (05.00.00. 8)

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3. Sadykov J.N., Ermatov Z.D. Development of technology for the production of coated surfacing electrodes for wear-resistant surfacing // International Journal Of Advanced Research in Science, Engineering and Technology ó India, 2021. ó Vol.8, 10 (November). ó pp. 18551 ó 18556 (05.00.00. 8)

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